

# Domain versatility in plant AB-toxins: Evidence for a local, pH-dependent rearrangement in the 2 $\gamma$ lectin site of the mistletoe lectin by applying ligand derivatives and modelling

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**Abstract** Mistletoe lectin is a potent biohazard. Lectin activity in the toxic dimer primarily originates from the 2 $\gamma$ -subdomain (Tyr-site) of the B-subunit. Crystallographic information on lectin–sugar complexes is available only at acidic pH, where lectin activity is low. Thus, we mapped ligand-binding properties including comparison to ricin's Tyr-site at neutral pH. Using these results and molecular dynamics simulations, a local conformational change was rendered likely. The obtained structural information is valuable for the design of potent inhibitors. © 2008 Federation of European Biochemical Societies. Published by Elsevier B.V. All rights reserved.

**Keywords:** Agglutinin; Lectin; Mistletoe; Ribosome-inactivating protein; Ricin

## 1. Introduction

Ribosome-inactivating plant lectins are potent biohazards and also attractive models to study evolutionary routes of intra- and interprotein divergence [1,2]. AB-type toxins combine a toxic A-subunit with the lectin part (B-subunit). Starting from a primordial 40-residues peptide, a series of gene duplications is supposed to have led to the common structural arrangement of the B-chain. It consists of two tandemly arrayed domains (1 and 2) established by four subdomains each ( $\alpha$ ,  $\beta$ ,  $\gamma$ ,  $\delta$ ) [3]. Carbohydrate-binding activity is found in the 1 $\alpha$  and 2 $\gamma$  subdomains. The potent toxin ricin, the role model of this class, harbours Trp37 in its low-affinity 1 $\alpha$  site and Tyr248 in its high-affinity 2 $\gamma$  site [3]. In solution it is monomeric, in contrast to the weakly toxic dimeric agglutinin of the castor bean. The galactoside-specific agglutinin from *Viscum album* L. (VAA; also called viscumin or ML-I) uniquely combines the activities as toxin/agglutinin in one protein [4]. Binding to cells and multivalent glycoproteins is of nM affinity and can engage the 1 $\alpha$  domain around Trp38 and the 2 $\gamma$  domain around Tyr249 [5–7]. Because dimer formation spatially

restricts accessibility to the Trp-site, the Tyr-site is central for glycan binding and toxicity above submicrogram/ml-levels [7].

Of note for structural design of inhibitors, crystals of VAA-sugar complexes had been grown in acidic pH, i.e. pH 3.4 and 2.5 [8,9], a condition detrimental for lectin activity [10]. The concern for pH-dependent local conformational changes in the binding site arises. It is also valid in the case of ricin, which shares this sensitivity toward pH decrease [11]. To address this issue and also enable interprotein comparison, we here present mapping of the ligand-binding properties of VAA's Tyr-site and, assisted by molecular dynamics simulations, develop a structural model of this site at neutral pH.

## 2. Materials and methods

### 2.1. Saccharides

The panel of saccharides was established from sources as given in detail previously [12,13].

### 2.2. Isolation and labelling of VAA

Purification, labelling with a biotin derivative and quality/activity controls were performed as described previously [5,7]. Biotin-labelled VAA was subsequently acetylated with *N*-acetylimidazole, in the absence or presence of 0.1 M lactose, using a 2-fold excess (w/w) of the reagent [5]. Protein concentration was determined by the Lowry assay using concanavalin A (Sigma) as standard [14]. Radioiodination of VAA using IODO-GEN (Pierce Eurochemie) up to a typical specific activity of  $15 \pm 5$   $\mu$ Ci/mg was carried out in the presence of 0.1 M lactose to protect the carbohydrate-binding sites from modification.

### 2.3. Quantitative binding studies

Plastic microwells were coated with 50  $\mu$ l of asialofetuin solution for 16 h at 4 °C and binding of <sup>125</sup>I-VAA to the wells was assayed essentially as described [12], except that the buffer used was 5 mM sodium phosphate buffer, pH 7.2, 0.2 M NaCl. For binding assays with biotinylated VAA, precoated wells were incubated with 50  $\mu$ l of lectin solution for 2 h at 20 °C, and the extent of bound lectin was monitored by measuring the amount of streptavidin associated to the wells after incubation for 1 h at 20 °C with 50  $\mu$ l (15000 cpm) of <sup>125</sup>I-streptavidin solution in the same buffer. Streptavidin (Sigma) was iodinated using IODO-GEN.

The affinity of VAA for the tested saccharides was estimated by determining the amount of <sup>125</sup>I-lectin bound to wells coated with 50  $\mu$ g/ml asialofetuin, after incubation at 20 °C with the <sup>125</sup>I-lectin solution in the absence or presence of different concentrations of the sugars.

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#### 2.4. Molecular dynamics (MD) simulations

Crystallographic parameters of the B chain (pdb entry 1oql) were used as starting point to adapt the Tyr-site residues to neutral pH. Crystallographic water molecules were retained, *N*-acetyl-D-glucosamine groups were removed, and hydrogen atoms were added with Sybyl [15]. The Amber 03 force field was used to model the protein, the generalized amber force field for galactose [16]. Partial charges on gal-

actose were calculated with the AM1-BCC method [17], using Antechamber in the AMBER 8 package [18]. Following addition of one chlorine ion, the system was solvated in an orthorhombic box of TIP3P water molecules ( $83.1 \times 86.8 \times 68.6$  Å).

Simulations were carried out with the AMBER 8 suite of programs [18]. The particle mesh Ewald method [19] was used to treat long-range electrostatic interactions; a non-bonded cut-off of 10 Å was used for

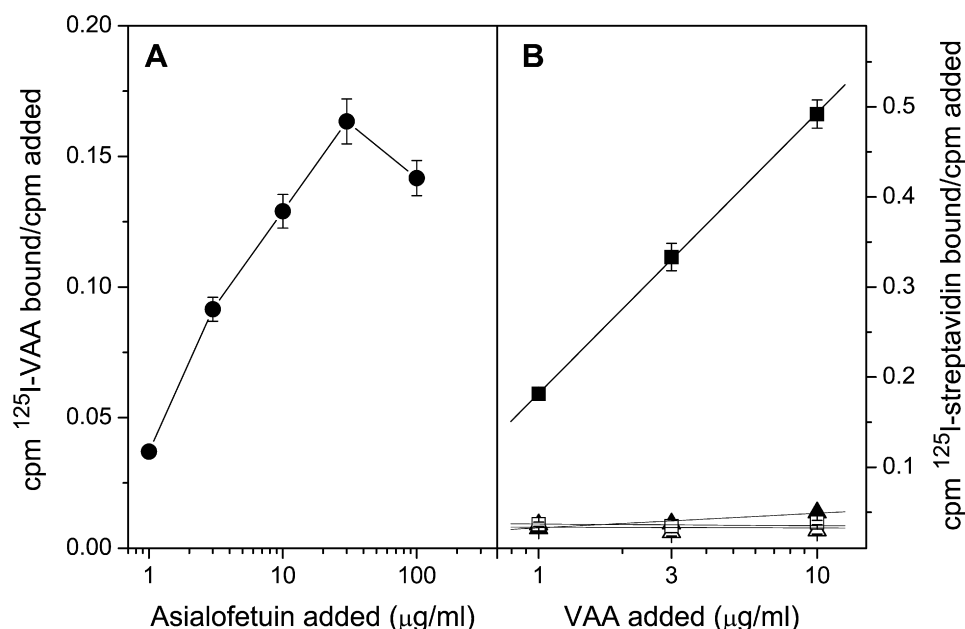


Fig. 1. Binding of VAA to asialofetuin-coated wells. (A) Binding of <sup>125</sup>I-VAA to wells coated with asialofetuin solutions of different concentrations was determined as described under Section 2. (B) Biotinylated VAA was treated with *N*-acetylimidazole in the absence (▲, △) or presence (■, □) of 0.1 M lactose. Microwell surfaces coated with asialofetuin (100 µg/ml) were then incubated, in the absence (filled symbols) or presence (open symbols) of 0.1 M lactose, with the acetylated biotinylated proteins at three different concentrations. The amount of bound lectin was finally assessed by measuring the binding of <sup>125</sup>I-streptavidin to the wells.

Table 1

Apparent association constants for the binding of methyl β-lactoside analogues to VAA

Unit derivatized	Compound	$K_a$ ( $M^{-1} \cdot 10^{-3}$ )		Relative $K_a$	
		VAA	Ricin	VAA	Ricin
	Methyl β-lactoside	$1.92 \pm 0.02$	$17.8 \pm 0.2$	1	1
<b>β-D-Galactopyranose</b>					
C-2'	2'-Deoxy	$0.23 \pm 0.02$	$3.1 \pm 0.1$	0.12	0.17
	2'-O-Methyl	$0.28 \pm 0.01$	$13.0 \pm 4.9$	0.14	0.73
C-3'	3'-Deoxy-3'-fluoro	<0.1	<0.1	<0.05	<0.005
	3'-O-Methyl	<0.1	$0.50 \pm 0.01$	<0.05	0.03
C-4'	4'-Deoxy	<0.1	$0.33 \pm 0.01$	<0.05	0.02
	4'-Deoxy-4'-fluoro	<0.1	<0.1	<0.05	<0.005
C-6'	4'-O-Methyl	<0.1	$0.61 \pm 0.01$	<0.05	0.03
	6'-Deoxy	$0.78 \pm 0.4$	$0.62 \pm 0.01$	0.4	0.03
	6'-Deoxy-6'-fluoro	$0.53 \pm 0.01$	$5.26 \pm 0.8$	0.3	0.29
	6'-O-Methyl	$0.50 \pm 0.05$	$10.3 \pm 0.3$	0.3	0.58
<b>β-D-Glucopyranose</b>					
C-1	1-Deoxy	$1.50 \pm 0.02$	$18.2 \pm 1.0$	0.78	1.0
C-2	2-Deoxy	$1.15 \pm 0.1$	$16.9 \pm 2.6$	0.59	0.95
	2-O-Methyl	$1.1 \pm 0.1$	$10.9 \pm 0.7$	0.57	0.61
C-3	3-Deoxy	$1.8 \pm 0.2$	$37.0 \pm 12$	0.93	2.1
	3-O-Methyl	$0.4 \pm 0.1$	$3.2 \pm 0.03$	0.20	0.18
	3-Deoxy-3-Methyl	$0.17 \pm 0.01$	$11.5 \pm 0.4$	0.1	0.64
	3-Epi	$0.34 \pm 0.02$	$2.6 \pm 0.1$	0.17	0.15
C-6	6-Deoxy	$1.30 \pm 0.05$	$8.8 \pm 0.3$	0.67	0.49
	6-O-Methyl	$1.1 \pm 0.1$	$7.7 \pm 0.3$	0.57	0.43

Relative  $K_a$  values were calculated taking  $K_a$  for methyl β-lactoside as unit. Values for ricin are from reference [11].

van der Waals interactions. A harmonic restraint of  $50 \text{ kcal mol}^{-1} \text{ \AA}^2$  was applied on protein residues, with the exception of binding-site residues (within  $6.5 \text{ \AA}$  of galactose) and galactose. The SHAKE algorithm [20] was applied to all bonds involving hydrogen atoms. After minimization, the system was gradually warmed to  $300 \text{ K}$  in  $250 \text{ ps}$ ; the simulation was then run for  $5 \text{ ns}$  in the NPT ensemble ( $T = 300 \text{ K}$ ;  $P = 1 \text{ atm}$ ), with an integration time-step of  $2.0 \text{ fs}$ .

### 3. Results

#### 3.1. Binding of VAA to asialofetuin

Tyrosine-radioiodinated VAA bound to asialofetuin-coated wells proportionally to the amount of asialofetuin added to the well (Fig. 1A). Carbohydrate-independent binding in the presence of  $0.1 \text{ M}$  lactose was always below  $1\%$ . The concentration of  $^{125}\text{I}$ -labelled lectin in the well was typically  $20 \mu\text{g/ml}$ , neatly above the  $3 \mu\text{g/ml}$ -limit for dimer dissociation becoming appreciable [4]. To attribute binding to the Tyr-site, biotinylated VAA was used. If tyrosine acetylation harms activity and this effect is neutralized by lactose, modification of Tyr249 will account for the loss of activity, as indeed shown in Fig. 1B. Binding of dimeric  $^{125}\text{I}$ -VAA to asialofetuin, as does the unlabelled lectin to asialofetuin-Sepharose 4B [7], is thus mediated by the Tyr-sites. They were next subjected to mapping in two stages at neutral pH.

#### 3.2. Carbohydrate-binding profile of the Tyr-sites

First, 20 galactose-based saccharides were tested. The apparent association constants were calculated from the plot of the reciprocal of the lectin fraction bound to the well versus the inhibitor concentration. Most disaccharides only slightly surpassed binding activity of galactose (Supplementary Material, Table). As control, the  $K_a$  estimated for the binding of lactose was very close to the value of  $(1.1 \pm 0.1) \times 10^3 \text{ M}^{-1}$  previously found by isothermal titration calorimetry at  $25^\circ\text{C}$  and assigned to the Tyr-sites [7].

#### 3.3. Chemical mapping with synthetic methyl $\beta$ -lactoside derivatives

To dissect the contribution of the hydroxyl groups, we performed a detailed chemical mapping analysis using a series of synthetic deoxy-, fluorodeoxy- and *O*-methyl derivatives of methyl  $\beta$ -lactoside (Table 1). As expected, the most important impact originated from changing groups at the galactose moiety. A large loss in affinity was observed for the derivatives at positions  $3'$  and  $4'$  of the galactose unit indicating that these hydroxyls are key groups in the recognition by VAA, as also observed for ricin. At position  $2'$  there is a weak but significant polar interaction, whereas the hydroxyl group at  $6'$  is only marginally involved in the binding. Of note, ricin tolerates *N*-acetylation of galactosamine at the  $2'$ -position. These results served as input to infer a pH-dependent conformational change.

#### 3.4. Molecular dynamics simulations

In the crystal structures of VAA in complex with galactose and lactose at acid pH [8,9], the strongest interactions at the Tyr-site involve the hydroxyl groups at positions  $3'$  and  $4'$ , but no involvement of HO- $2'$  is visible. To reconcile experimental with structural data, MD simulations were run at neutral pH. In the crystal structure, the side chain of Lys254 is

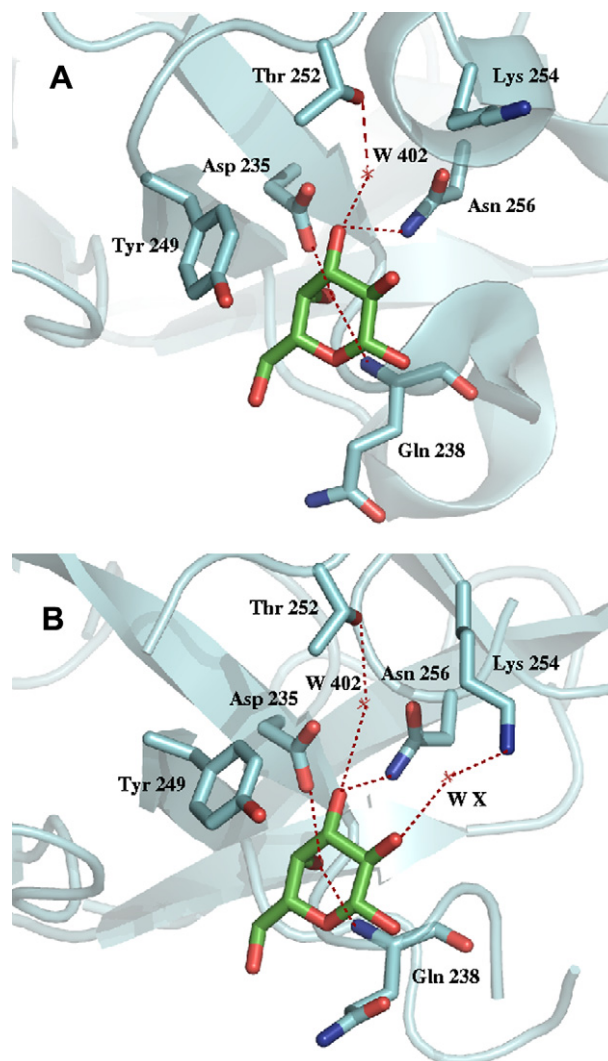


Fig. 2. pH Dependence of the mode of binding of galactose to the Tyr-site of VAA. (A) Binding mode in the crystal structure of the VAA-galactose complex (pdb 1oql) and (B) after a MD simulation at neutral pH. Galactose is shown in green, protein residues in cyan and hydrogen bonds as dotted red lines.

fully oriented towards the solvent (Fig. 2A). After a simulation time of  $500 \text{ ps}$ , folding of Lys254 towards galactose was observed and a water molecule, W X, was found to mediate the interaction between the C-2 hydroxyl of galactose and the  $\text{N}^Z$  group of lysine (Fig. 2B). The vicinity of Lys254 to position 2 of the galactose ring can explain why bulky substituents at this position, as in  $2'$ -*O*-methyl  $\beta$ -lactoside and, more significantly, *N*-acetylgalactosamine, will not enter easily the Tyr-site.

### 4. Discussion

The results presented here reveal pH-dependent conformational changes and highlight differences in the mode of binding at the Tyr-sites of VAA and ricin. The presented interaction profiles of ligands with substitutions at positions  $2'$  and  $6'$  of galactose mainly contribute to explain this disparity, manifested also on the level of cells [21]. The illustrated

pH-dependent change in VAA provides a structural basis for excluding bulky substituents at the 2'-position. Looking at the involvement of HO-6', it is strongly dependent on the pH in ricin, its role diminishing at acidic pH. In the ricin-lactose complex, crystallized at pH 4.75 [3,22], the C-6' hydroxyl makes no contacts with the protein. At neutral pH, however, the formation of a salt bridge between Arg236 and Glu199 favours the establishment of a H-bond between N<sup>E</sup> (Arg236) and the 6'-hydroxyl group. MD runs revealed the presence of the contacts but they were not stable throughout the simulation, as similarly reported recently [23]. In VAA's Tyr-site, Arg236 and Glu199 are replaced by Ala and Ser, respectively, precluding such a pH-dependent switch for VAA, which tolerates  $\alpha$ 2,6-sialylation well compared to ricin [24,25]. On the grounds of minor effects of extension to disaccharides in most cases in solid-phase and cell assays [10,26], the noted change at the OH-2' position of galactose can thus account for modulating ligand affinity at this level of structural complexity.

In summary, the chemical mapping delineates alterations in interaction profiles between VAA and ricin and gives reason to postulate a local pH-dependent conformational change for VAA inferred by modelling. Its detection will optimize design of potent inhibitors, the described strategy being valuable beyond this special protein.

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## Appendix A. Supplementary material

Supplementary data associated with this article can be found, in the online version, at [doi:10.1016/j.febslet.2008.05.035](https://doi.org/10.1016/j.febslet.2008.05.035).

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